Impurities In Biomolecules Institute for International Research

## Monitoring & Predicting Biomolecular Aggregation Using Light Scattering

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## Outline

- Light Scattering Technologies
- Batch Light Scattering Applications
  - Kevin Mattison Malvern Instruments
- Flow Mode Light Scattering Applications
  - Ewa Folta-Stogniew Yale University
- Closing
- Appendix



## **Biomolecular Stability**

#### With regard to aggregation:

- In the absence of stabilizing "conditions", when small particles collide, London forces can dominate the interaction, leading to particle aggregation.
- In order to stabilize a formulation against aggregation, particle collisions must be minimized. This can be accomplished using:
- Electrostatic Effects wherein the presence of charge leads to a repulsive force between the particles.
- Steric Effects wherein the presence of adsorbed or attached additives (known as chaotropic agents) prohibit particles from getting close enough together for London forces to dominate.



## Stabilizing Effects

In the absence of steric effects, the stability of a system to aggregation is determined by the balance of repulsive and attractive forces which the particles experience as they approach one another. The rule of thumb for electrostatically stable suspension is +/- 30 mV for the zeta potential.



In the absence of electrostatic effects, steric stability is enhanced upon addition of chaotropic or "structure disrupting" agents, which reduce the likelihood of particles getting close enough for London forces to take over.





Unstable

Stable

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## **Evidence Of Instability?**

#### Insulin formulations at t = 0 and 12 months





## Common Approach – Time Trials

According to sizing results, the sample is completely aggregated within 7 days of preparation.





## But Can We Predict?

**Light Scattering:** Low energy photon induces an oscillating dipole in the electron cloud. As the dipole oscillates, energy is radiated in all directions. This radiated energy is called "scattered light."

Rayleigh Scattering Profile





## Static Light Scattering (SLS)

Average scattering intensity leads to the particle molecular weight, 2nd virial coefficient, and radius of gyration ( $R_q$ ).





## Dynamic Light Scattering (DLS)

Correlation of short time scale ( $\mu$ s) intensity fluctuations gives the diffusion coefficient, hydrodynamic size, polydispersity, and particle size distribution.



q = Scattering vector $R_H = Radius$ T = Temperature

- D = Diffusion coefficient
- k = Boltzmann constant
- $\eta = Solvent viscosity$



## Electrophoretic Light Scattering (ELS)

Measured parameter is the frequency shift of the scattered light.



The frequency shift is proportional to the electrophoretic mobility, which is a function of the particle surface potential. Hence ELS gives us information regarding the charge on the particle.

 $\mu = K \left( \frac{\Delta \upsilon}{E} \right)$ 



## Why Light Scattering?

- The scattering intensity:
- varies with the mass and concentration according to the Rayleigh Expression
- ▶ is proportional to
  - M<sub>w</sub>
  - M<sub>n</sub><sup>2</sup>
  - R<sup>6</sup>
- is non-invasive
- is ideal for aggregate detection & quantification in low volume, low concentration biological samples.





Peak	D <sub>l</sub> (nm)	%	<b>%M</b>
1	5.95	5	93
2	46.0	94	6



## Lysozyme - Comparison Of Radii



M = 14.5 kDa

 $V_p$  = 0.73 mL/g

$R_g =$	1.4	17	nm
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For	R <sub>g</sub>
Sphere	= 0.774 R <sub>H</sub>
Coil	= 0.816 R <sub>H</sub>
Cylinder	= 1.732 R <sub>H</sub>



## What Is M<sub>W</sub>?

 $\ensuremath{\mathsf{M}_{\mathsf{W}}}$  is the mass or weight average molecular weight.

Number Average

Weight Average



 $N_i$  = the number of particles in each weight class  $M_i$  = the molecular weight of each weight class  $m_i$  = the mass of particles in each weight class

As a consequence,  $M_W$  is more heavily weighted by larger particles in the sample.

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## Light Scattering Applications

## **Batch Mode Measurements**















## **Monitoring Stability**





## **Predicting Stability**



Formulation B



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## **Prediction & Observation**





## Zeta Potential - Shelf Life Correlation

Formulations with varying soy fiber content.





## Electrostatic vs. Steric Stability

Polymer adsorption to cationic liposomes in PBS reduces the electrostatic while enhancing the steric stabilization. All are stable.



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## Salt Effects On Aggregation

Electrostatic shielding enhances BSA aggregation for NaCl concentrations  $\geq$  500 mM at the isoionic point. But the aggregation is reversible, suggesting that it is "non-denaturing".





## Polydispersity (Pd) From DLS

Pd is representative of the particle size distribution width, with high polydispersity being indicative of oligomerization and/or aggregation.





## **High Concentration Sizing**





## Using $T_M$ As A Stability Predictor



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## **Melting Point - Shelf Life Correlation**

Samples prepared in phosphate buffered saline at 1.0 mg/mL protein concentration.





## Using A<sub>2</sub> As A Stability Predictor

#### A<sub>2</sub> is closely correlated with sample solubility





## Batch Mode Challenges – Oligomers!

- Static LS
  - Classical
- elevated weight average Molar Mass (M<sub>w</sub> weight average)
- angle dependent intensity

- Dynamic LS
  - Quasi-elastic
- autocorrelation function cannot be fit to single exponential (Cumulant)
- high polydispersity (%Pd > 15%)

#### Missing information: how much and/or what size?

#### **Solutions**

- Sample fractionation followed by batch measurements
- Column separation with simultaneous LS characterization

## **Light Scattering Applications**

## Flow Mode Measurements









## **Three Detector monitoring**



## Ovalbumin 43 kDa

88% monomer

8% dimer

1.5% trimer

3% aggregates < 1MDa

0.4% 1-100 MDa



Intensity of scattered light  $\sim M_w^*c$ 

due to their high Mw aggregates scatter very strongly

A monomeric protein 43 kDa and aggregates 10 MDa at 2 mg/mL:



#### Molar mass distribution for multiple analyses

Ovalbumin 43 kDa

#### automated template processing of five data sets



#### **Determination of Weight Fractions**





#### Differences in population based on molar mass distribution



#### Differences in population based on molar mass distribution



Differences in population based on molar mass distribution

#### Ovalbumin 43 kDa

Oligomeric state	Average Mw ± SD [kDa] (5 analyses) Mw = 108 ± 17	Average Mw ± SD [kDa] (3 analyses) Mw = 141 ± 3	Fraction of Mass [% of total] (5 analyses) Mw = 108 ± 17	Fraction of Mass [% of total] (3 analyses) Mw = 141 ± 3
Mono (20-50 kDa)	43.0 ± 0.1	42.80 ± 0.02	88.1 ± 0.1	85.23 ± 0.06
Di (50-96 kDa)	82.7 ± 0.4	84.1± 0.2	7.68 ± 0.04	9.4 ± 0.0
Tri (96-130 kDa)	114 ± 4	121.8 ± 0.7	1.54 ± 0.05	1.9 ± 0.0
Agg. (0.13 –1 MDa)	270 ±10	284 ± 2	2.18 ± 0.08	2.87± 0.06
Agg. (1 –100 MDa)	10±1 x10 <sup>3</sup>	10.9±0.4 x10 <sup>3</sup>	0.4 ± 0.0	0.6 ± 0.0

# Morphology of aggregates from angular dependence of LS signal; size determination- Rg



# Inferring conformational information from the relationship between molecular size (Rg) and molecular weight (Molar Mass)

$$R_g \sim M^v$$

log(R<sub>g</sub>) versus log(MM)

Slope = v

For	ν
Sphere	0.33
Coil	0.5
Rod	1

Rollings, J.E. (1992) in *"Laser Light Scattering in Biochemistry",* Eds. S.E. Harding, D. B. Sattelle and V. A. Bloomfield; p. 275-293



## Shape analysis: log(Rg) versus log(MM)

#### Aggregates of Ovalbumin vs. "amyloid-type" fibers





#### Shape analysis: shape factor $\rho = R_g/R_h$





<sup>B</sup> Rizos, A. K., Spandidos, D. A., and Krambovitis, E., (2003) Int. J. Med., 12, 559-563



#### Shape analysis: shape factor $\rho = R_g/R_h$



# Determination of the oligomeric state of modified protein from SEC/LS analysis

- 1. Glycosylated proteins
- 2. Proteins conjugated with polyethylene glycol
- 3. Membrane protein present as a complex with lipids and detergents

#### Input:

- Polypeptide sequence
- Chemical nature of the modifier

#### Results:

- Oligomeric state of the polypeptide
- Extend of modification (grams of modifier /gram of polypeptide)

#### "three detector method"

## **Three Detector Method**

$$MW_p = \frac{k^*(LS)(UV)}{\varepsilon(RI)^2}$$

$MW_p$	Molecular Weight (polypeptide)
3	extinction coefficient
LS	light scattering intensity
UV	absorbance (ε)
RI	refractive index change
k	calibration constant

Yutaro Hayashi, Hideo Matsui and Toshio Takagi (1989) Methods Enzymol,172:514-28 Jie Wen, Tsutomu Arakawa and John S. Philo (1996) Anal Biochem, 240(2):155-66

## MWp = 91.39 x [(LS)\*(UV)/( $\epsilon$ \*(RI^2))]

Protein	MW (kDa)
Ova	43
BSA(1)	66
BSA(2)	132
Ald	156
Apo-Fer	475



#### PEG-ylated protein: 75 kDa

36 kDa polypeptide + 39 kDa PEG



PEG-ylated protein: 75 kDa

36 kDa polypeptide + 39 kDa PEG



#### PEG-ylated oligo: 48.3 kDa

#### 8.3 kDa oligo + 40 kDa PEG



PEG (40K) MM = 41.0 kDa

Polydispersity= 1.001

40K PEG + 8.3 kDa oligo

PEG-oligo MM = 48.5 kDa

#### Protein "F"



Protein "F" frictional ratio  $R_h/R_s = 1.85$  non-spherical shape



Protein: 12 kDa; WT and three mutants

Interaction with the column effects



## **Capabilities**

#### Static LS

- fast and accurate determination of molar masses (weight average)
  - glycosylated protein, conjugated with PEG, protein-lipids-detergent complexes, protein-nucleic acid complexes
- accuracy of ± 5% in Molar Mass determination
- easy to implement, fully automated (data collection and data analysis)
- highly reproducible (no operator bias)
- SEC/MALS excellent in detecting and quantifying population with various oligomeric state in protein

#### **Dynamic LS**

- very fast detection of aggregates
- great dynamic range
- well suited to study kinetics of aggregation
- DLS detector available in a plate reader format for high volume analyses

#### Combined data about MM, Rg and Rh - shape information (multiangle static and dynamic LS)

- via frictional ratio Rh/Rs
- via shape factor v, from log(Rg) vs. log(MM) plot
- via shape factor  $\rho,$  from Rg/Rh ratio

## Limitations

#### Static LS

- measures weight average molar mass needs fractionation to resolve different oligomeric states
- possible losses of sample during filtration and fractionation
- Imitation on solvent choices (related to a fractionation step)
- SEC/SLS/DLS dilution during experiment

#### **Dynamic LS**

- measures hydrodynamic radius, which is affected by shape
- cannot discriminate between shape effects and changes in oligomeric states, *i.e.* non-spherical shape mimics oligomerization
- needs fractionation to resolve low number oligomers when present in mixture

## **Common Light Scattering Specifications**

Parameter	Specification		
Size range - DLS	0.6 nm to 6 um Diam		
Size range - Zeta Potential	10 nm to 20 um Diam		
Concentration range	0.1 mg/mL (Lys) to 30w%		
Minimum sample volume	2 uL		
Temperature control	-4 to 130 °C		
Accessories			
Polarization filters for rotational correlation measurements			
Wavelength filters for fluorescing samples			
Automatic titrators			
Cross-correlation configurations			
Plate readers for high throughput applications			

Multi-angle configurations for full MW & Rg range

Flow cells for HPLC applications

#### Ken Williams

Director of W.M. Keck Biotechnology Resource Laboratory at Yale University School of Medicine

#### NIH

#### Users of SEC/LS Service

#### http://info.med.yale.edu/wmkeck/biophysics



# **Questions?**

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